# Fluid & Its Properties

**<u>Fluid:</u>** Matter which can flow (Liquid & Gas).

Internal Resisting shear force (Shear stress) is zero at rest.

**<u>Density</u>** ( $\rho$ ): Density ( $\rho$ ) =  $\frac{Mass(m)}{Volume(V)}$ 

 $\rho_{Air} = 1.2 \text{ kg/m}^3$   $\rho_{Water} = 1000 \text{ kg/m}^3$   $\rho_{Hg} = 13600 \text{ kg/m}^3$ 

Specific weight (w): Specific weight (w) =  $\frac{\text{Weight (W)}}{\text{Volume (V)}}$ w =  $\rho$  g

$$w_{H_20} = 9810 \text{ N/m}^3$$

**Specific Gravity (S)**: Specific Gravity (S) =  $\frac{\text{Density of fluid}}{\text{Density of standard fluid}}$ 

Standard fluid is Water for liquid

Standard fluid is Air for Gas

<u>Compressibility (β):</u> variation of volume with change in pressure.

$$\beta = \frac{-dV/V}{dP}$$

**<u>Bulk modulus (K)</u>**:  $K = \frac{dP}{-dV/V} = \frac{dP}{d\rho/\rho} = \frac{1}{\beta}$ 

Isothermal Bulk Modulus of Ideal Gas  $(K_I) = P$ Adiabatic Bulk Modulus of Ideal Gas  $(K_A) = \gamma P$ 

Ratio of Specific heats

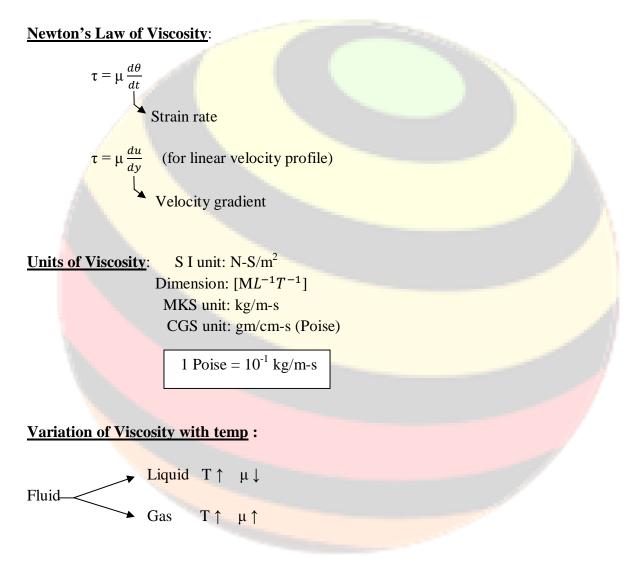
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## GATE REVOLUTION

No Slip Condition: Relative velocity of fluid wrt a solid surface in zero at solid surface.

<u>Viscosity ( $\mu$ )</u>: When fluid is at motion the resistance offered by one layer of fluid to the adjacent layer of fluid in known as viscosity.



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### GATE REVOLUTION

<u>**Kinematic Viscosity (**</u> $\upsilon$ ): Kinematic Viscosity ( $\upsilon$ ) =  $\frac{\text{Dynamic Viscosity (}{\mu})}{\text{Density (}{\rho})}$ 

Units of Kinematic Viscosity: MKS unit: m<sup>2</sup>/sec

Dimension:  $[M^0L^2T^{-1}]$ 

CGS unit: cm<sup>2</sup>/sec (Stoke)

1 Stoke =  $10^{-4}$  m<sup>2</sup>/sec

Newtonian Fluid: Fluids which follows the Newton's law of Viscosity.

e.g. Air, Water, Oil etc.

 $\mu_{water} = 1 \text{ cP at } 20^{\circ} \text{ C}$ 

**Ideal fluid:** Fluid having zero viscosity ( $\mu = 0$ ).

Cohesion & Adhesion:

 $\begin{array}{c} A \\ A \\ A \\ \end{array} \begin{array}{c} \leftrightarrow \\ B \\ \end{array} \end{array}$ 

(Cohesion: force of Attraction between same molecules)

(Adhesion: force of Attraction between different molecules)

<u>Surface tension</u> ( $\sigma$ ): Tensile force normal to the line drawn on the surface & acts along the plane of surface

Surface Tension ( $\sigma$ ) =  $\frac{\text{Tensile force (F)}}{\text{length (L)}}$ 

 $\sigma$  depends on both liquid & gas which are forming interface

$$\sigma_{\rm H_2O/Air} = 0.0736 \, \rm N/m$$

**<u>Variation of surface tension with temp</u>**:  $T \uparrow \sigma \downarrow$ 

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## GATE REVOLUTION

#### **<u>Gauge Pressure under various conditions</u> :**

- $\Delta P = \frac{4\sigma}{D}$  (for water Droplet & Air Bubble in water)
- $\Delta P = \frac{8\sigma}{D}$  (For soap bubble)
- $\Delta \mathbf{P} = \frac{2\sigma}{D} \text{ (for liquid jet)}$

Where D is the diameter

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